



MUTAH UNIVERSITY  
Faculty of Engineering  
Department of Chemical Engineering



## Industrial Physical Chemistry

### COURSE SYLLABUS

Course Code	Course Name	Credits	Contact Hours
0404213	Industrial Physical Chemistry	3	

#### INSTRUCTOR/COORDINATOR

Name	Prof. Adnan Al-Harabsheh
Email	Adnan@mutah.edu.jo
Website	

#### TEXTBOOK

Peter Atkins, "**Physical Chemistry**" 8 Edition, Oxford press 2006

#### Other Supplemental Materials

PPT notes

#### SPECIFIC COURSE INFORMATION

##### A. Brief Description of the Content of the Course (Catalog Description)

Physical chemistry is a combined science of physics, chemistry, thermodynamics, electrochemistry, and quantum mechanics. It functions to provide molecular-level interpretations of observed macroscopic phenomena. Typically, changes in temperature, pressure, volume, heat, and work of systems in the solid, liquid, and or gas phase are correlated to microscopic atomic and molecular interactions.

##### B. Pre-requisites (P) : 0404112

##### C. Course Type (Required or Elective)

Required (Compulsory department course)

**SPECIFIC GOALS****A. Specific Outcomes of Instruction**

The main theme of this course is to introduce the main physical chemistry concepts that will lay the foundation for the chemical Engineering students to understand thermodynamics in later courses. Some Chemical engineering applications will be studied in this course.

**B. Student Outcomes Addressed by the Course**

1	2	3	4	5	6	7				
✓										

**BRIEF LIST OF TOPICS TO BE COVERED**

List of Topics	No. of Weeks	Contact Hours
Introduction to physical chemistry	1-2	6
Perfect gas	3	3
Real gas	4	3
Energy and the first law of thermodynamics	5-6	6
Work & Heat	7	6
Enthalpy	8-9	6
Thermochemistry	10	3
Second law (Reaction spontaneity and the direction of thermodynamic change)	11-12	6
Phase Transformation of pure substance	13	3
Phase Diagrams	14-16	6
<b>Total</b>	<b>16</b>	<b>48</b>

**METHODS OF ASSESSMENT**

No.	Method of assessment	Week and Date	%
1	Home works , quizzes . discipline	Through semester	<b>30</b>
2	Mid Exam	Middle of semester	<b>30</b>
3	Final examination	End of semester	<b>40</b>
<b>Total</b>			<b>100</b>

